

Notes

Asymmetric Synthesis of New Chiral Cyclic Ketols

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Introduction

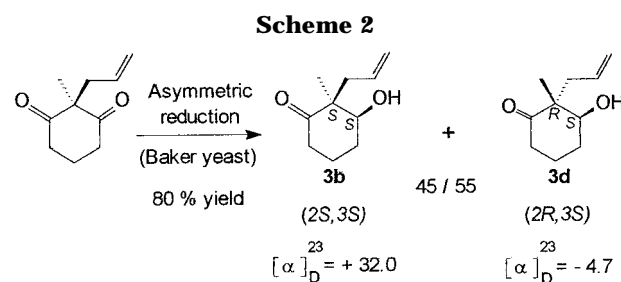
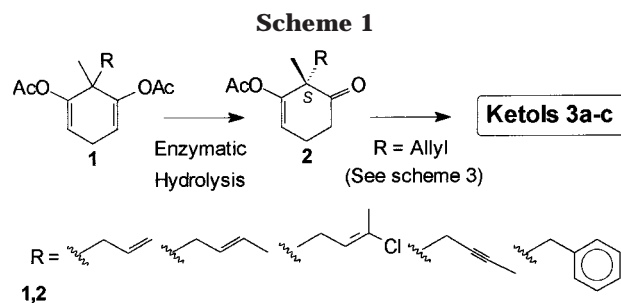
In our previous work^{1,2} we reported an asymmetric enantioselective synthesis of keto enol acetates **2** (>98% ee) bearing a quaternary stereogenic center by enzymatic hydrolysis of prochiral dienol diacetates **1**. In this paper, we now demonstrate that starting from these chiral materials **2**, the synthesis of new enantiopure compounds can be achieved with a total control of the absolute configuration. To test the potential of these new synthons, we chose to prepare stereoisomeric cyclic ketols **3a–c** from keto enol acetate **2a** (Schemes 1 and 3). This kind of ketol has been used for the synthesis of bioactive compounds such as Zoapatanol.³

In the literature chiral six-membered ring ketols are prepared by microorganism asymmetric reduction of the corresponding prochiral diketones. This reaction leads to a mixture of two diastereoisomeric alcohols, and ratios depend on diketone substitution. Ketols **3b** and **3d**, with an (*S*) configuration on the stereogenic center bearing the hydroxy group, have been prepared using this methodology (Scheme 2).⁴

Results and Discussion

In our preliminary studies, we synthesized ketol **3a** (*2R,3R*) to assign the (*S*) absolute configuration to the quaternary stereogenic center of keto acetate **2a**, which had been obtained by enzymatic hydrolysis.² Our approach relied on the NaBH₄ reduction of **2a** followed by the cleavage of the enol ester function with methyllithium and purification to afford ketol **3a** (*R,R*) in 40% overall yield (Scheme 3).

We have now been able to prepare the (*S,S*) enantiomer (ketol **3b**) from the same keto acetate **2a** (Scheme 3). Dioxolane **4** was indeed obtained in good yield from keto



acetate **2a**, and subsequent cleavage of the enol ester group by reaction with potassium *tert*-butoxide⁵ provided ketone **5**. Alcohol **6b** was then obtained using LiBHET₃ as reducing agent.⁶ The major diastereoisomer (1*S*,2*S*/1*R*,2*S*: 96/4) was isolated by chromatography in 91% yield. Compound **6b** was finally converted to ketol **3b** by treatment with 3 N HCl.

For the preparation of ketol **3c** (*2S,3R*), diastereoselectivity of carbonyl reduction of the ketone **5** (Scheme 3) has to be inverted. We chose to apply the Luche procedure⁷ using a lanthanide chloride derivative as chelating agent. Treatment of ketone **5** with europium trichloride⁸ and sodium borohydride in methanol provided a diastereoisomeric mixture out of which alcohol **6c**, presenting (1*R*,2*S*) configuration, was isolated. Hydrolysis with 3 N HCl afforded the ketol **3c**. The configurations of **3a–c** prepared as shown in Scheme 3 were attributed according to literature values⁴ for **3b** and **3d**.

In conclusion, enantiopure ketols **3a,c** have been easily synthesized from keto enol acetate **2a** obtained by enzymatic hydrolysis of prochiral dienol diacetate **1a**. Ketol **3c** has never before been prepared to our knowledge. Our methodology which allows the preparation of ketols such as **3a** and **3c**, with an (*R*) configuration for the stereogenic center bearing the hydroxy group, is complementary of the asymmetric reduction of diketones⁴ which affords the alcoholic group with an (*S*) configuration. This work is currently being developed on the synthesis of linear enantiomerically pure ketols by transformation of cyclic chiral molecules.

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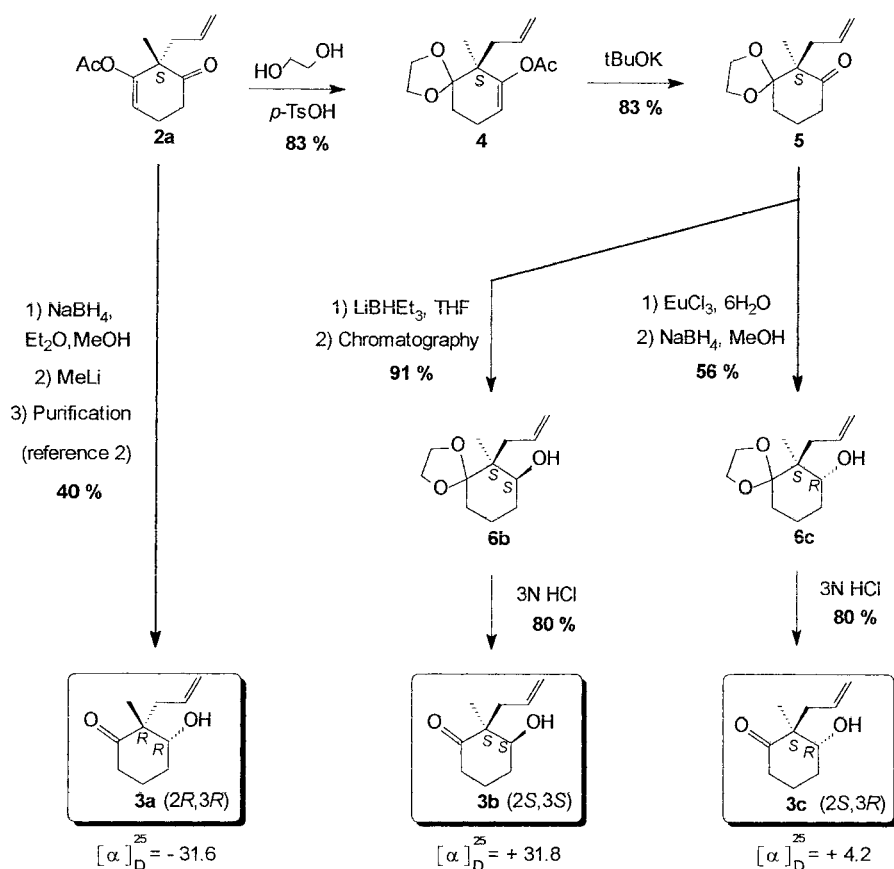
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Scheme 3



Furthermore, it is noteworthy that this method can be used as a good tool to produce six-membered cyclic compounds with three stereogenic centers by suitable modifications of the carbonyl moiety of **3a–c**.

Experimental Section

^1H and ^{13}C NMR spectra were recorded on a spectrometer operating at 200 MHz for ^1H and 50 MHz for ^{13}C from CDCl_3 solutions unless otherwise noted. J values are given in hertz. IR spectra were obtained with a FT-IR spectrophotometer. Mass spectra were recorded on a mass spectrometer (EI: electronic impact; CI: chemical ionization with $t\text{BuH}$). GLC analysis were performed on a gas chromatograph, using an HP-1 (5 m \times 0.53 mm \times 2.65 μm film) column for GC routine, HP-1 (25 m \times 0.25 mm) column for diastereomers separation. Flash chromatography was performed on (230–400 mesh ASTM) silica gel, unless otherwise noted, with petroleum ether (distillation temp $<$ 60 $^\circ\text{C}$)–diethyl ether mixtures as eluent. The progress of reactions was monitored by GC or by TLC (Et_2O –petroleum ether). Microanalyses were performed by the INSA Laboratoires, Rouen.

(*S*)-1-Acetoxy-3,3-(ethylenedioxy)-2-methyl-2-(prop-2-enyl)cyclohex-1-ene (4). Keto acetate **2a** (105 mg, 0.5 mmol) was added to a solution of ethylene glycol (0.5 mL, 9 mmol) and p -TsOH (5 mg, 0.03 mmol) in benzene (10 mL) and the mixture refluxed for 2 h in a Dean–Stark apparatus. Saturated aqueous NaHCO_3 was added. Aqueous layer was extracted with pentane. Combined organic layers were washed with brine, dried over MgSO_4 and concentrated under reduced pressure. Flash chromatography (Et_2O /petroleum ether: 15/100) on silica gel (60 \AA , $>$ 440 mesh) afforded the dioxolane **4** (106 mg, 83%). ^1H NMR δ_{H} (200 MHz) 1.05 (s,3H), 1.69–1.80 (m,2H), 2.08 (s,3H), 2.30 (dt, J 7.2, J 1.2, 2H), 3.92–3.97 (m,4H), 4.90–5.01 (m,2H) and 5.37 (t, J 4.0, 1H); ^1H NMR δ_{H} (200 MHz, C_6D_6) 1.28 (s,3H), 1.46–1.72 (m,2H), 1.70 (s,3H), 2.01–2.14 (m,2H), 2.55 (dt, J 7.3, J 1.3, 2H), 3.35–3.50 (m,4H), 5.00–5.17 (m,2H), 5.50 (t, J 4.0, 1H), 6.09–6.30 (m,1H); ^{13}C NMR δ_{C} (50 MHz) 18.4, 20.7, 21.1, 26.9, 40.6, 44.6, 64.6, 64.7, 111.4, 113.9, 115.5, 136.2, 149.5,

169.2; IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$ 1760, 1640; MS m/z (EI) 55, 69, 73, 86, 99, 109, 124, 169, 210, 252; $[\alpha]_D^{25}$ ($c = 1, \text{CH}_2\text{Cl}_2$).

(*S*)-3,3-(Ethylenedioxy)-2-methyl-2-(prop-2-enyl)cyclohexan-1-one (5). A solution of keto acetate **4** (330 mg, 1.3 mmol) in THF (2.5 mL) was added at -20 $^\circ\text{C}$ to a suspension of potassium *tert*-butoxide (162 mg, 1.43 mmol) in THF (2.5 mL). The mixture was stirred for 45 min after which 1 N hydrochloric acid and diethyl ether (10 mL) were added to it. The aqueous layer was extracted with Et_2O , and the combined organic layers were washed with brine, dried over MgSO_4 , concentrated, chromatographed (Et_2O /petroleum ether: 10/100/silica gel: 60 \AA , $>$ 440 mesh) to afford ketone **5** (230 mg, 83%); ^1H NMR δ_{H} (200 MHz) 1.22 (s, 3H), 1.37–1.78 (m, 4H), 2.00–2.25 (m, 2H), 2.36–2.63 (m, 2H), 3.32–3.37 (m, 4H), 4.92–5.04 (m, 1H); ^{13}C NMR* δ_{C} (50 MHz) 14.3, 19.3, 29.6, 37.2, 39.3, 58.8, 63.1, 65.3, 113.4, 117.5, 133.5, 211.4; IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$ 1715, 1640; MS m/z (CI, $t\text{BuH}$) 99, 211; $[\alpha]_D^{25}$ ($c = 1, \text{CH}_2\text{Cl}_2$). Anal. Calcd for $\text{C}_{12}\text{H}_{18}\text{O}_3$: C, 68.55; H, 8.63. Found: C, 68.71; H, 8.63; mp 48 $^\circ\text{C}$.

(1*S*,2*S*)-3-(1,3-Dioxolan-2-yl)-2-methyl-2-(prop-2-enyl)cyclohexan-1-ol (6b). To a solution of ketone **5** (210 mg, 1.0 mmol) in THF (5 mL) was added dropwise at -78 $^\circ\text{C}$ a solution of lithium triethylborohydride in THF (1 M in THF, 1.1 mL, 1.1 mmol). The mixture was stirred overnight. Methanol (6 mL) and water (6 mL) were added at -78 $^\circ\text{C}$, and the mixture was stirred for 30 min. The suspension was concentrated in vacuo, and the residue was dissolved in Et_2O and washed with 1 N hydrochloric acid (acidic pH). The layers were separated, and the aqueous layer was extracted with Et_2O . The combined organic layers were washed with brine, dried over MgSO_4 , and evaporated to afford (210 mg) alcohol **6** in a 96/4 1*S*,2*S*/1*R*,2*S* diastereomeric ratio. The major diastereomer (1*S*,2*S*)-**6b** was obtained by flash chromatography (Et_2O /petroleum ether: 5/100) on silica gel (60 \AA , $>$ 440 mesh) (195 mg, 91%); ^1H NMR δ_{H} (200 MHz) 0.99 (s, 3H), 1.48–1.76 (m, 6H), 2.00 (dd, J 13.3, J 8.2, 1H), 3.14 (t, J 10.2, OH), 3.60 (dt, J 10.2, J 3.1, 1H), 3.84–4.00 (m,4H), 5.01–5.13 (m,2H) and 5.78–5.96 (m,1H); ^{13}C NMR δ_{C} (50 MHz) 17.7, 19.0, 28.0, 30.1, 35.1, 44.9, 64.0, 65.3, 75.5, 112.9, 117.5 and

135.0; IR (neat) ν_{\max} /cm⁻¹ 3450, 1640; MS m/z (CI) 195, 213; m/z (EI) 55, 69, 86, 99, 108, 122, 139, 150, 194, 212.

(1*R*,2*S*)-3-(1,3-Dioxolan-2-yl)-2-methyl-2-(prop-2-enyl)cyclohexan-1-ol (6c). To a solution of ketone **5** (200 mg, 0.95 mmol) in methanol (5 mL) was added at -78 °C europium trichloride (520 mg, 1.4 mmol), and the mixture was stirred for 1 h. Sodium borohydride (36 mg, 0.95 mmol) was added, and the mixture was stirred overnight. Saturated aqueous NaHCO₃ (5 mL) was added at -78 °C and then Et₂O (5 mL) at room temperature. The layers were separated, and the aqueous layer was extracted with Et₂O. The combined organic layers were washed with brine, dried over MgSO₄, and evaporated to afford (200 mg) alcohol **6** in a 44/56 1*S*,2*S*/1*R*,2*S* diastereomeric ratio. The major diastereomer (1*R*,2*S*)-**6c** was obtained by flash chromatography (Et₂O/petroleum ether: 5/100) on silica gel (60 Å, >440 mesh) (112 mg, 56%); ¹H NMR δ_{H} (200 MHz) 1.00 (s, 3H), 1.49–1.72 (m, 6H), 2.25–2.32 (m, 2H), 2.84 (OH), 3.65–6.72 (m, 1H), 3.85–3.97 (m, 4H), 4.97–5.10 (m, 2H), 5.81–6.02 (m, 1H); ¹³C NMR δ_{C} (50 MHz) 15.1, 18.1, 28.1, 29.7, 39.4, 46.0, 64.4, 64.8, 74.0, 113.0, 116.8, 135.8; IR (neat) ν_{\max} /cm⁻¹ 3540 and 1640; MS m/z (CI) 195 and 213; m/z (EI) 55, 69, 86, 99, 108, 122, 139, 150, 194, 212.

Ketols 3b and 3c. Solutions of the alcohols **6b** or **6c** (120 mg, 0.6 mmol) in 3 N hydrochloric acid (4 mL) were stirred overnight at room temperature. The solution was extracted with CH₂Cl₂ (4 × 2 mL), washed with brine, dried over MgSO₄, and concentrated. The crude product was purified by flash chromatography (Et₂O/petroleum ether: 20/100) to afford respectively ketols **3b** or **3c** (80 mg, 80%).

(2*S*,3*S*)-3-Hydroxy-2-methyl-2-(prop-2-enyl)cyclohexan-1-one (3b). ¹H NMR δ_{H} (200 MHz) 1.13 (s,3H), 1.60–2.11 (m, 4H), 2.32–2.44 (m, 4H), 2.70 (s,OH), 3.57 (dd, *J* 7.0, *J* 3.3, H), 5.02–5.13 (m,2H), 5.62–5.83 (m,1H); ¹³C NMR δ_{C} (50 MHz) 19.5, 20.5, 28.4, 36.4, 37.6, 54.1, 76.3, 117.7, 133.6, 214.5; IR (neat) ν_{\max} /cm⁻¹ 3480, 1700, 1640; m/z (CI) 151, 169; MS m/z (EI) 55, 67, 81, 93, 109, 122, 135, 150, 168. Anal. Calcd for C₁₀H₁₆O₂: C, 71.39; H, 9.59, Found: C, 71.53; H, 9.38; [α]²³_D (*c* = 0.48, CHCl₃), for comparison Literature⁴: (2*S*,3*S*) **3b**: xxxxx (*c* = 0.42, CHCl₃), (2*R*,3*S*) **3d**: [α]²³_D (*c* = 0.6, CHCl₃).

(2*S*,3*R*)-3-Hydroxy-2-methyl-2-(prop-2-enyl)cyclohexan-1-one (3c). Yield: 80 mg, 80%; ¹H NMR δ_{H} (200 MHz) 1.11 (s, 3H), 1.58–2.07 (m,4H), 2.25–2.45 (m, 4H), 2.82 (s,OH), 3.85 (dd, *J* 7.3, *J* 3.0, H), 5.03–5.12 (m,2H), 5.63–5.84 (m,1H); ¹³C NMR δ_{C} (50 MHz) 17.6, 20.3, 28.3, 37.6, 40.0, 54.3, 74.9, 117.8, 133.8, 214.2; IR (neat) ν_{\max} /cm⁻¹ 3520, 1705, 1645; m/z (CI) 151 and 169; MS m/z (EI) 55, 67, 81, 93, 109, 122, 135, 150, 168. Anal. Calcd for C₁₀H₁₆O₂: C, 71.39; H, 9.59, Found: C, 71.53; H, 9.38; [α]²⁵_D (*c* = 0.5, CHCl₃). For literature comparison, (2*S*,3*S*) **3b**: [α]²³_D (*c* = 0.42, CHCl₃); (2*R*,3*S*) **3d**: [α]²³_D (*c* = 0.6, CHCl₃).

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